



Convenient synthesis of [3]catenane by olefin metathesis dimerizations

Hajime Iwamoto,^a Koji Itoh,^a Hiroyuki Nagamiya^b and Yoshimasa Fukazawa^{a,*}

^aDepartment of Chemistry, Graduate School of Science, Hiroshima University, Higashi-Hiroshima 739-8526, Japan

^bDepartment of Applied Chemistry, Graduate School of Science and Engineering, Tokyo Institute of Technology, 2-12-1 Ookayama, Meguro, Tokyo 152-8552, Japan

Received 25 April 2003; revised 5 June 2003; accepted 9 June 2003

Dedicated to Emeritus Professor Soichi Misumi on the occasion of his 77th birthday

Abstract—[3]Catenane **5a** and **5b** were synthesized conveniently by olefin metathesis dimerization of pseudorotaxanes **3a** and **3b**. The yields of **5a** and **5b** were influenced by concentrations of **3a** and **3b**, and a ring size of a center wheel of [3]catenane. © 2003 Elsevier Ltd. All rights reserved.

Topological molecules, catenanes, rotaxanes, knots¹ constitute a major research field in supramolecular chemistry. These molecules display interesting physical properties, such as photoinduced intramolecular electron transfer, electrochemically triggered molecular motions, and photochemical dethreading process.^{1e,2,3} A number of these multicomponent systems have been named ‘molecular machines.’ Many research groups synthesized these molecules, and interesting molecular motions were found.^{1e,4} Olefin metathesis has been extensively utilized for the synthesis of macrocycles, carbocycles and heterocycles.⁵ Several groups showed ring-closing metathesis (RCM) was extremely useful for catenanes, rotaxanes and knots syntheses.⁶ Recently Smith and co-workers reported the C₂-symmetric cyclophane skeletons by olefin metathesis dimerization.⁷ Now we wish to report convenient synthesis of [3]catenane by using olefin metathesis dimerization.

Our synthetic plan of [3]catenane is outlined in Figure 1. [3]Catenanes were synthesized by several procedures.⁸ However, to the best of our knowledge, there is no previous report on the synthesis of [3]catenane by olefin metathesis dimerizations. Formation of pseudorotaxane from a wheel and an axis molecule, followed by olefin metathesis dimerization should afford [3]catenane. Dimerization process consists of intermolecular olefin metathesis and ring-closing metathesis, which should occur competitively. Therefore, there is a possibility to obtain other products, such as [2]catenane, highly complicated catenanes and oligomeric compounds. In order to obtain [3]catenane efficiently, it is necessary that each step, formation of pseudorotaxane, intermolecular olefin metathesis, ring-closure olefin metathesis, were performed effectively. These steps are affected by concentrations of wheel and axis molecules. Intermolecular olefin metathesis prefers

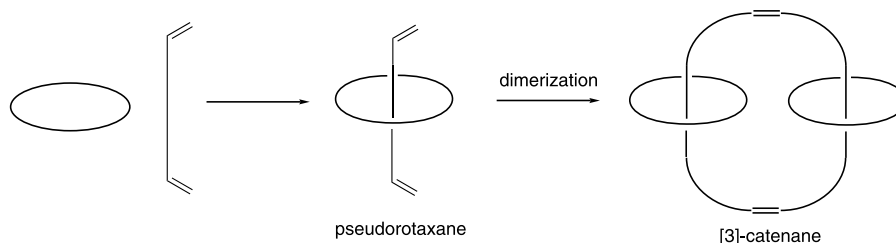
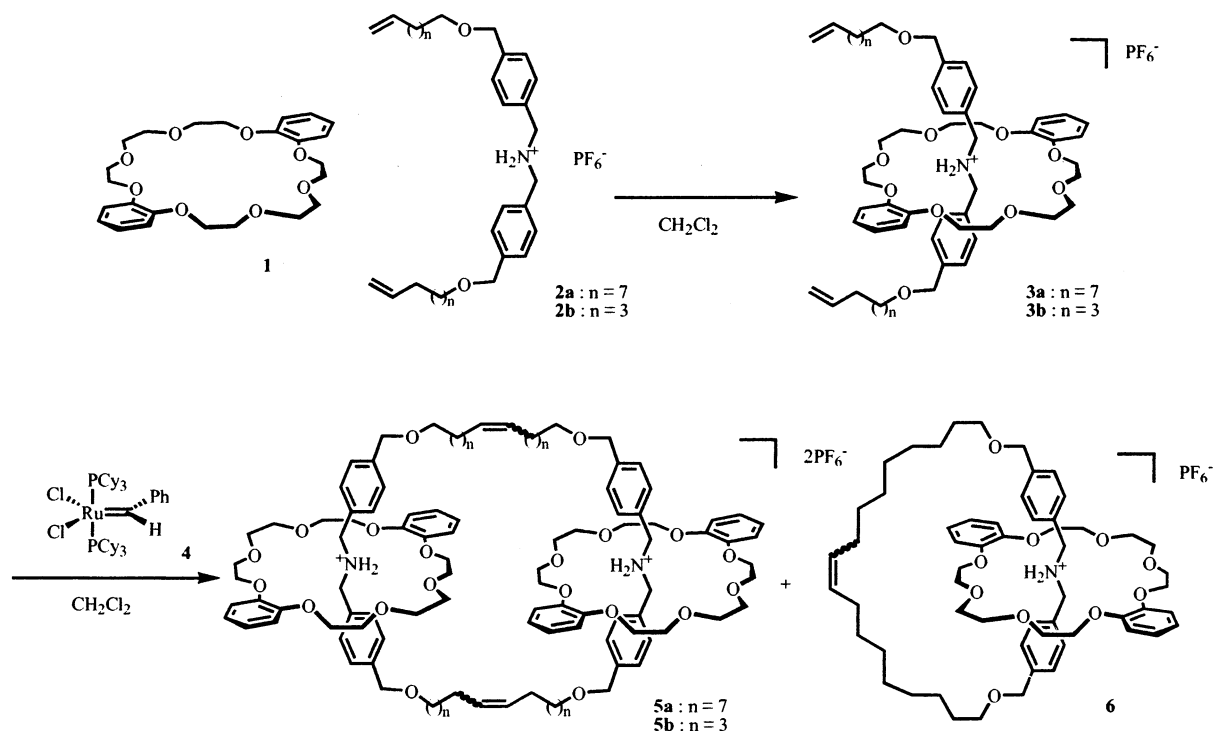


Figure 1. Outline of [3]catenane synthesis.

Keywords: ring-closing metathesis; olefin metathesis dimerization; [3]catenane.

* Corresponding author. Tel.: +81-824-24-7427; fax: +81-824-24-0724; e-mail: fukazawa@sci.hiroshima-u.ac.jp



Scheme 1. Synthesis of [3]catenane using olefin metathesis.

highly concentrated condition; on the other hand RCM prefers high dilution condition. Also highly concentrated condition is needed for the formation of pseudorotaxane.

The synthesis of [3]catenane was performed as shown in Scheme 1. Treatment of pseudorotaxane **3a**, which was derived from dibenzo[24]crown-8 (DB24C) **1** and an equivalent of ammonium salt having two terminal olefin **2a**⁹ in dichloromethane, with a catalytic amount of ruthenium carbene complex **4**¹⁰ gave the desired catenation products **5a**,⁶¹¹ and oligomeric by-products. When **2b**, which had shorter alkyl chains than **2a**, was used instead of **2a**, only [3]catenane **5b** and by-products were obtained, but [2]catenane could not be found. These compounds were purified by column chromatography, GPC and HPLC analysis. ¹H NMR, and ESI-TOF mass spectrum¹² of the synthesized [2] and [3]catenanes were shown in Figures 2 and 3, respectively. These ¹H NMR show the signals of disubstituted olefin proton, which occurred olefin metathesis and crown ether and ammonium salt component. The positive-ion ESI-TOF mass gave the correspondent exact mass of each product and observed isotopic distributions of the products were similar to theoretical ones.

The yields of [3]catenane and [2]catenane are influenced by concentrations of dibenzo[24]crown-8 (DB24C) **1** and ammonium salt having two terminal olefin **2a** (Table 1). When concentration of **1** and **2a** was 0.01 M of dichloromethane solution, [2]catenane **6** was obtained in 56% yield, and [3]catenane **5a** was only in 2% yield. Increasing the concentrations, the yield of [3]catenane **5a** increased, but that of [2]catenane **6** was

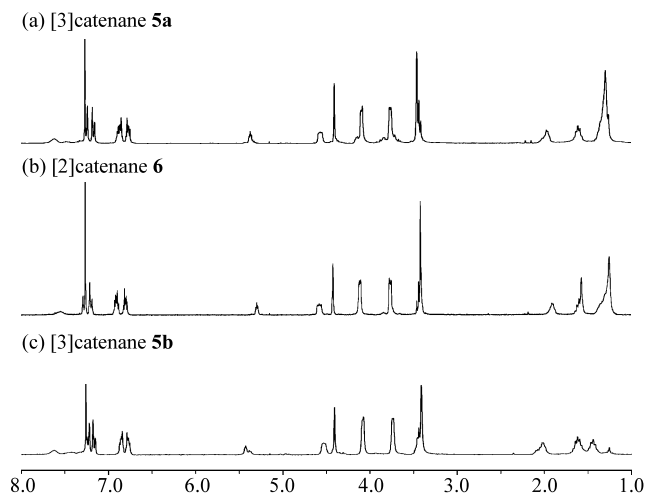


Figure 2. ¹H NMR spectrum of (a) [3]catenane **5a**; (b) [2]catenane **6** and (c) [3]catenane **5b**.

decreased. On the other hand, in the case of **2b**, only [3]catenane **5b** was obtained under several conditions of concentration.¹³ Apparently, under high dilution condition using **2a**, ring-closing metathesis preferred to dimerization process, and then [2]catenane must be major product. As concentration of **1** and **2a** was increased, intermolecular olefin metathesis was easier to occur and the yield of [3]catenane was increased. **2b** has the axle, not long enough to form [2]catenane, therefore ring-closing metathesis would be difficult to react and the dimerization process would take priority over other processes.

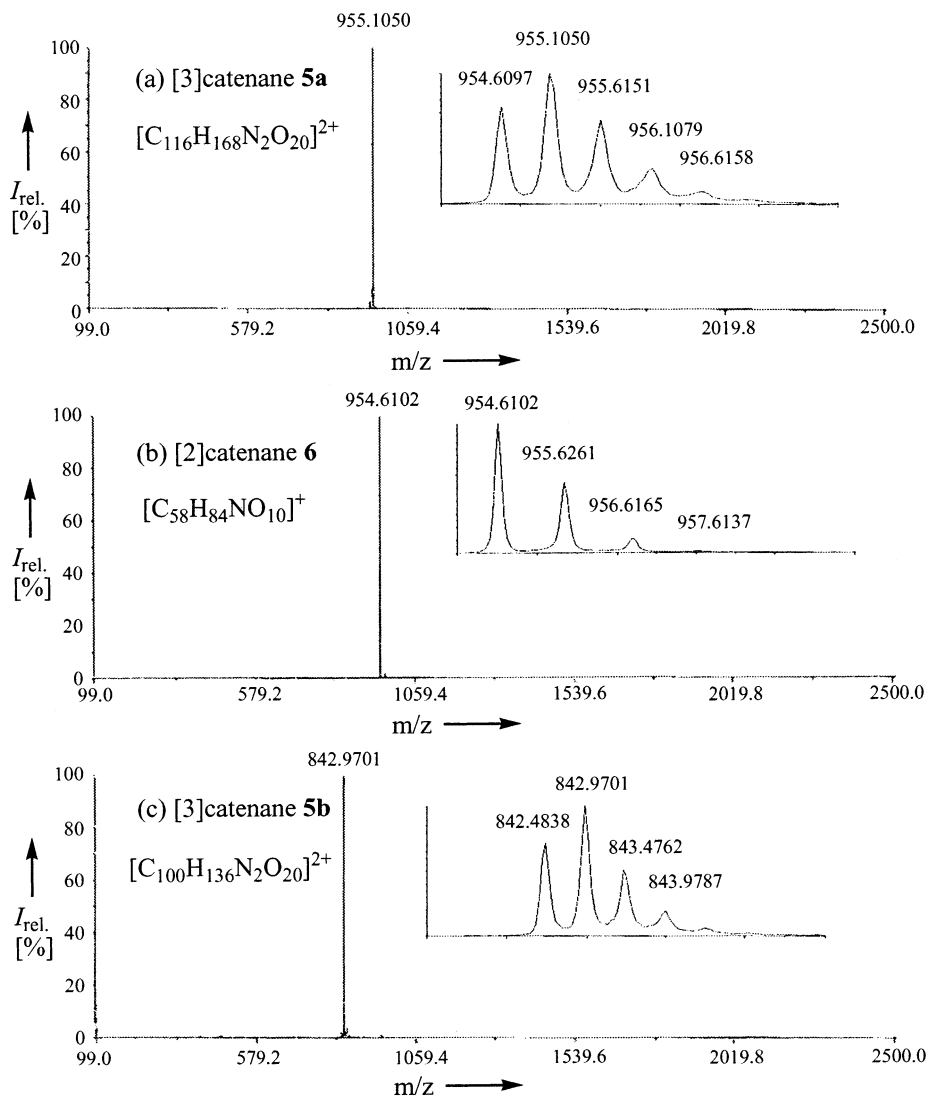


Figure 3. ESI-TOF mass spectrum of (a) [3]catenane **5a**; (b) [2]catenane **6** and (c) [3]catenane **5b**.

Table 1. Yields of [3]catenanes **5a** and **5b**, and [2]catenane **6**

Alkyl chain length n	Concentration [M] ^a	Yield	
		[3]Catenane (%)	[2]Catenane (%)
7 (2a)	0.002	2	56
	0.01	16	41
	0.02	20	31
3 (2b)	0.002	25	0
	0.01	33	0
	0.02	35	0

^a Both of dibenzo[24]crown-8 (DB24C) **1** and ammonium salt **2a** and **2b** were prepared as the same concentration.

In conclusion, we have developed a convenient method of synthesis of [3]catenane by olefin metathesis dimerization. The method was affected by a concentration of the substrates, and the size of center wheel of [3]catenane made by olefin metathesis dimerization.

Acknowledgements

This work was supported by Grant-in-Aid for Scientific Research (No. 10304053) from the Ministry of Education, Science, Sports and Culture, Japan, which is gratefully acknowledged.

References

1. (a) Dietrich-Buchecker, C. O.; Sauvage, J.-P. *Chem. Rev.* **1987**, *87*, 795–810; (b) Sauvage, J.-P. *Acc. Chem. Res.* **1990**, *23*, 319–327; (c) Amabilino, D. A.; Stoddart, J. F. *Chem. Rev.* **1995**, *95*, 2725–2828; (d) Philp, D.; Stoddart, J. F. *Angew. Chem., Int. Ed. Engl.* **1996**, *35*, 1154–1196; (e) Balzani, V.; Gómez-lópez, M.; Stoddart, J. F. *Acc. Chem. Res.* **1998**, *31*, 405–414.
2. See review: Sauvage, J.-P. *Acc. Chem. Res.* **1998**, *31*, 611–619.
3. (a) Murakami, H.; Kawabuchi, A.; Kotoo, K.; Kinitake, M.; Nakashima, N. *J. Am. Chem. Soc.* **1997**, *119*, 7605–7606; (b) Kawaguchi, Y.; Harada, A. *Org. Lett.* **2000**, *2*, 1353–1356; (c) Lane, A. S.; Leigh, D. A.; Murphy, A. J. *J. Am. Chem. Soc.* **1997**, *119*, 11092–11093; (d) Benniston, A. C.; Harriman, A. *Angew. Chem., Int. Ed. Engl.* **1993**, *32*, 1459–1461; (e) Armaroli, N.; Balzani, V.; Collin, J.-P.; Gavina, P.; Sauvage, J.-P.; Ventura, B. *J. Am. Chem. Soc.* **1999**, *121*, 4397–4408; (f) Collier, C. P.; Mattersteig, G.; Wong, E. W.; Luo, Y.; Beverly, K.; Sapaio, J.; Raymo, F. M.; Stoddart, J. F.; Heath, J. R. *Science* **2000**, *289*, 1172–1175; (g) Brouwer, A. M.; Frochot, C.; Gatti, F. G.; Leigh, D. A.; Mottier, L.; Paolucci, F.; Roffia, S.; Hurlpel, G. W. H. *Science* **2001**, *291*, 2124–2128.
4. (a) Balzani, V.; Credi, A.; Raymo, F. M.; Stoddart, J. F. *Angew. Chem., Int. Ed.* **2000**, *39*, 3348–3391; (b) Jiménez, M. C.; Dietrich-Buchecker, C.; Sauvage, J.-P. *Angew. Chem., Int. Ed.* **2000**, *39*, 3284–3287; (c) Collier, C. P.; Mattersteig, G.; Wong, E. W.; Luo, Y.; Beverly, K.; Sampaio, J.; Raymo, F. M.; Stoddart, J. F.; Heath, J. R. *Science* **2000**, *289*, 1172–1175; (d) Watanabe, N.; Kihara, N.; Takata, T. *Org. Lett.* **2001**, *3*, 3519–3522; (e) Brouwer, A. M.; Frochot, C.; Gatti, F. G.; Leigh, D. A.; Mottier, L.; Paolucci, F.; Roffia, S.; Wurlpel, G. W. H. *Science* **2001**, *291*, 2124–2128; (f) Elizarov, A. M.; Chiu, S.-H.; Stoddart, J. F. *J. Org. Chem.* **2002**, *67*, 9175–9181; (g) Ros, T. D.; Guldi, D. M.; Morales, A. F.; Leigh, D. A.; Prato, M.; Turco, R. *Org. Lett.* **2003**, *5*, 689–691.
5. (a) Lee, C. W.; Choi, T.-L.; Grubbs, R. H. *J. Am. Chem. Soc.* **2002**, *124*, 3224–3225; (b) Gao, Y.; Wei, C.-Q.; Burke, T. R., Jr. *Org. Lett.* **2001**, *3*, 1617–1620; (c) Trost, B. M.; Chisholm, J. D.; Wroblewski, S. T.; Jung, M. J. *J. Am. Chem. Soc.* **2002**, *124*, 12420–12421.
6. (a) Wolovsky, R. *J. Am. Chem. Soc.* **1970**, *92*, 2132–2133; (b) Ben-Efraim, A. A.; Batich, C.; Wasserman, E. *J. Am. Chem. Soc.* **1970**, *92*, 2133–2135; (c) Gruter, G. M.; Akkerman, O. S.; Bickelhaupt, F. *Tetrahedron* **1996**, *52*, 2565–2572; (d) Mohr, B.; Weck, M.; Sauvage, J. P.; Grubbs, R. H. *Angew. Chem., Int. Ed. Engl.* **1997**, *36*, 1308–1310; (e) Hamilton, D. G.; Sanders, J. K. M. *Chem. Commun.* **1998**, 1749–1750; (f) Rapenne, G.; Dietrich-Buchecker, C.; Sauvage, J.-P. *J. Am. Chem. Soc.* **1999**, *121*, 993–1001; (g) Kidd, T. J.; Leigh, D. A.; Wilson, A. J. *J. Am. Chem. Soc.* **1999**, *121*, 1599–1600; (h) Dietrich-Buchecker, C.; Sauvage, J.-P. *Chem. Commun.* **1999**, 615–616; (i) Weck, M.; Mohr, B.; Sauvage, J.-P.; Grubbs, R. H. *J. Org. Chem.* **1999**, *64*, 5463–5471; (j) Leigh, D. A.; Lusby, P. J.; Teat, S. J.; Wilson, A. J.; Wong, J. K. Y. *Angew. Chem., Int. Ed.* **2001**, *40*, 1538–1543; (k) Mobian, P.; Kern, J.-M.; Sauvage, J.-P. *J. Am. Chem. Soc.* **2003**, *125*, 2016–2017; (l) Wisner, J. A.; Beer, P. D.; Drew, M. G. B.; Sambrook, M. R. *J. Am. Chem. Soc.* **2002**, *124*, 12469–12476; (m) Coumans, R. G. E.; Elemans, J. A. A. W.; Thordarson, P.; Nolte, R. J. M.; Rowan, A. E. *Angew. Chem., Int. Ed.* **2003**, *42*, 650–654.
7. Smith, A. B., III; Kozmin, S. A.; Adams, C. M.; Paone, D. V. *J. Am. Chem. Soc.* **2000**, *122*, 4984–4985.
8. (a) Sauvage, J.-P.; Weiss, J. *J. Am. Chem. Soc.* **1985**, *107*, 6108–6111; (b) Dietrich-Buchecker, C. O.; Khemiss, A.; Sauvage, J.-P. *J. Chem. Soc., Chem. Commun.* **1986**, 1376–1378; (c) Kern, J.-M.; Sauvage, J.-P.; Weidmann, J.-L. *Tetrahedron* **1996**, *52*, 10921–10934; (d) Amabilino, D. B.; Ashton, P. R.; Stoddart, J. F.; White, A. J. P.; Williams, D. J. *Chem. Eur. J.* **1998**, *4*, 460–468; (e) Safarowsky, O.; Vogel, E.; Vögtle, F. *Eur. J. Org. Chem.* **2000**, 499–505; (f) Ashton, P. R.; Baldoni, V.; Balzani, V.; Claessens, C. G.; Credi, A.; Hoffmann, H. D. A.; Raymo, F. M.; Stoddart, J. F.; Venturi, M.; White, A. J. P.; Williams, D. J. *Eur. J. Org. Chem.* **2000**, 1121–1130; (g) Cabezon, K. B.; Cao, J.; Raymo, F. M.; Stoddart, J. F.; White, A. J. P.; Williams, D. J. *Chem. Eur. J.* **2000**, *6*, 2262–2273; (h) Chiu, A.-H.; Elizarov, A. M.; Glink, P. T.; Stoddart, J. F. *Org. Lett.* **2002**, *4*, 3561–3564.
9. Compound **2** was prepared by the method reported in the following paper; Ashton, P. R.; Glink, P. T.; Stoddart, J. F.; Tasker, P. A.; White, A. J. P.; Williams, D. J. *Chem. Eur. J.* **1996**, *2*, 729–736.
10. Schwab, P.; Fance, M. B.; Ziller, J. W.; Grubbs, R. H. *Angew. Chem., Int. Ed. Engl.* **1995**, *34*, 2039–2041.
11. Compounds **5a,b**, **6** were obtained as mixtures of isomers *E* and *Z* olefin, which could not be separated. Oligomeric by-products were not purified, therefore not identified.
12. ESI-TOF mass were carried out on a AppliedBioSystems coupled to a Mariner TK-3500 BioSpectrometry Workstation. The samples were dissolved in 1% AcOH in MeCN/1% AcOH in H₂O=50/50, and Bradykinin, Angiotensin I and Neurotensin were used as internal standards for obtaining high-resolution mass spectral data.
13. The higher concentrations more than 0.02 M did not give sufficient yields of [3] and [2]catenane, and oligomeric by-products were increased.